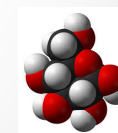
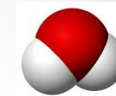
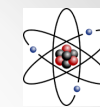


# Atoms, Molecules, Elements, Compounds and Mixtures

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Year 8 Chemical Sciences

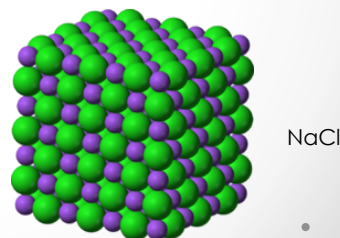
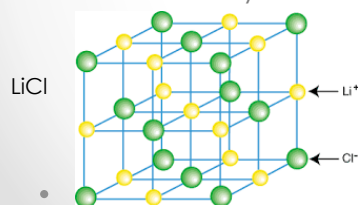
## Recap

- Atom
  - Smallest particle of matter and a building block
  - Nucleus-protons and neutrons, electrons
- Molecule
  - Clusters of atoms (building blocks) that can be the same or different
- Element
  - Substances made up of ONE type of atom
  - Each element has a unique set of properties
- Mixture
  - Molecules that make up the mixture are not identical
  - Components can be physically separated
  - Anything with list of molecules or compounds



## Compounds

- Chemical bonding
  - Cannot be separated
- Properties of compounds are **different** to the elements that make them up
- Molecules
  - Small with few atoms
  - Large with thousands of atoms
- Can form crystal lattices



## Activity 9

- Completing the Table

Substance	Appearance	Magnetism
Iron		
Sulfur		
Iron & Sulfur Mixture		
Iron & Sulfur after Heating		

## Iron and Sulfur: Appearance



Chunky, yellow powder



Fine, grey powder



Light grey fine powder with some yellow chunks

## Iron and Sulfur: Magnetism



No



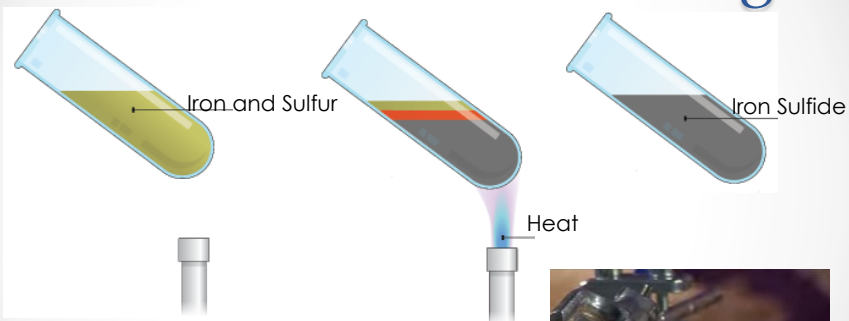
Yes



Iron filings 'stand' up, sulfur does not

Is it a Mixture? Why?

## Iron and Sulfur: Heating



<https://www.youtube.com/watch?v=A5H6DVe5FAI>

## Lump of Iron Sulfide



Black solid, sometimes in lumps or powder

Magnetic or NOT?

Is it a Mixture or Compound? Why?

## Word equation

Iron + sulfur  $\longrightarrow$  Iron sulfide

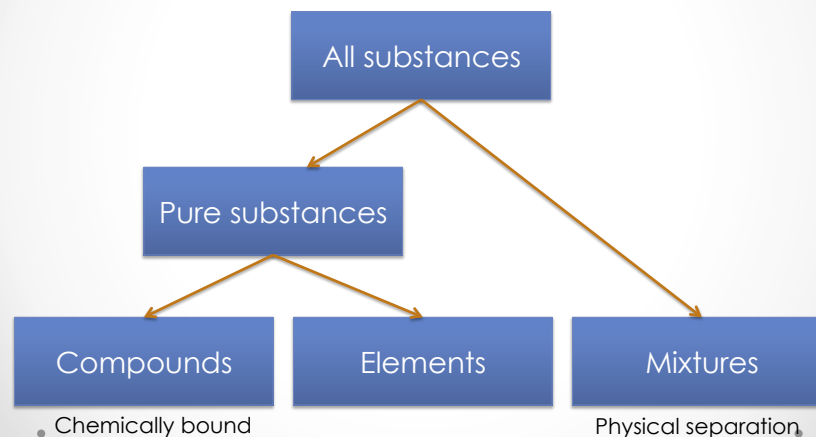


Fe + S



FeS

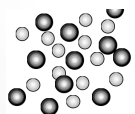
## How it fits together



## Questions

- Answer questions on Page 18 of Chemical Sciences Book

- What changes, if any, have occurred to the properties of the mixture heated in the test tube?
- How easy would it be to separate the iron and sulfur again? What has happened to make this task much harder?
- Write a word equation for the chemical reaction you observed. Your teacher will show you how to use an arrow in the equation.
- Explain the difference between a mixture and a compound.
- Which of these diagrams represents a mixture and which represents a compound?



● = iron atom  
● = sulfur atom

## Resources on Moodle

Reaction of iron with sulfur

<https://www.youtube.com/watch?v=A5H6DVe5FAI>

Atoms, elements and molecules

<http://splash.abc.net.au/home#!/media/1520179/atoms-elements-and-molecules>

Development of the periodic table

<http://splash.abc.net.au/home#!/media/106960/development-of-the-periodic-table>

The Elements: Find the elements-Tests difference between a compound and an element

<http://www.scootle.edu.au/ec/viewing/L2550/index.html>